Chapter 2

Details of chemicals, materials and equipments

In this chapter, details of chemicals and list of materials used for the synthesis of nanomaterials and details of instruments that used for the characterization of materials were discussed.

2.1.Materials

All the chemicals used were of analytical grade. Solvents used were of spectroscopic grade. Precursor substrate such as oil was procured commercially. List of other plant based precursors are include in **Table 2.1** while details are described under 'Materials' section of the respective chapters.

2.2. Details of Instruments

2.2.1. UV-visible study

UV-visible spectra were recorded on a Shimadzu 1601 PC UV-visible spectrophotometer.

2.2.2. FT-IR study

FT-IR spectrawere recorded on aPerkin Elmer spectrum 2 spectrophotometer using KBr pellets.

2.2.3. Raman study

Raman Spectra were recorded on a RenishawRM1000B LRM using a 514.5 nm($E_{laser} = 2.41 \text{ eV}$)Ar⁺ laser excitation source.

2.2.4. Powder X-ray diffraction study

Powder X-ray diffraction (XRD) measurements were carried out on a Philips X'PERT powder X-ray diffractometer with Cu-K α radiation (λ =1.54056 Å) with a scan speed 2^O/min at room temperature and on a Bruker AXS D8-Advance powder X-ray diffractometer with Cu-K α radiation (λ =1.5418Å) with a scan speed 2^O/min at room temperature.

2.2.5. Scanning electron microscopy

Scanning electron microscopy (SEM) images were obtained on a JEOL, JSM-6360 equipment, LEO, 1430vpequipment and on Quanta 150 equipmentwith an accelerating voltage of 1KV-30KV.

2.2.6. Transmission electron microscopy

Transmission electron microscopy (TEM) images were obtained on a JEOL, 9JSM-100CX equipment with an accelerating voltage of 60-200 KV. The sample powders were dispersed in ethanol, under sonication and TEM grids were prepared using a few of the dispersion followed by drying in air.

2.2.7. Energy dispersive X-ray study

Energy dispersive X-ray (EDS) pattern was recorded on JEOL, JED2300 SEM equipment, JEOL, 9JSM-100CX TEMequipment and Quanta 150 SEM equipment.

2.2.8. Tapdensity

The tapdensity was measured by mechanically tapping the synthesized material in a graduated vessel. The initial powder volume or mass is taken and then the vessel is mechanically tapped, and volume or mass readings are taken until no further volume or mass change is observed. The mechanical tapping is achieved by raising the cylinder or vessel and allowing it to drop, under its own mass. The volume of the graduated vessel was calibrated with distilled water.

2.2.9. Capacitance study

The capacitance measurements were carried out using cyclic voltammetry and galvanostatic charge/discharge method using a Potentiostat/Galvanostat (Gamry instruments, Reference 3000, Version 5.61) system.

2.2.10. Antioxidant activity

The antioxidant property was monitored by diphenyl picrylhydrazyl(DPPH) scavenging method. UV-visible spectrometer (Shimadzu 1800 PC) was used to determine the variation in DPPH concentrations at 517nm. A modified DPPH method was used to determine antioxidant activities of the synthesized material because materials are insoluble in methanol. The free radical scavenging assay was calculated by sonicating the mixture of synthesized materials (2, 4, 6, 8 and 15mg) and DPPH (3 ml, 100 μ M) in methanol. After centrifugation, absorbance is measured at 517 nm at an interval of 15, 30, 45 and 60 minutes.

2.2.11. Antimicrobial activity

The disc diffusion method was used to determine the inhibition zones. Sterile molten Mueller-Hinton Agar cooled to 45° C was inoculated with different organisms. The inoculums used were the young cultures and the inoculum size was standardized in such a way that each mL contains 108 cells. Using an aseptic technique, the inoculum was uniformly inoculated over the molten agar with sterile cotton swabs. A Whatman No 2 filter paper disc of 6 mm diameter containing 200 µL/disc of sample was placed over the inoculated medium. The plates were allowed to remain at room temperature for two hours. Then the plates were incubated at 37°C for 24 hours. The zone of inhibition was measured using a Zone Reader scale.

2.2.12. Photocatalytic activity

Photocatalytic activity of the material was studied by methylene blue (MB) degradation in visible light using 18 Watt CFL bulb in a closed chamber. The degradation efficiency was monitored using a Spectrophotometer, Systronics106.

2.2.13. Photoluminescence study

Photoluminescence spectrum was recorded on a Shimadzu RF-5301 PC spectrofluorophotometer at room temperature.

Sl. No.	Precursors
1.	Turpentine oil
2.	Sesame oil
3.	Soybean oil
4.	Sun flower oil
5.	Refined Mustard oil
6.	Palmolein oil
7.	Ghee(clarified butter from cow's milk)
8.	D-fructose
9.	1-Butanol
10.	Vinyl alcohol
11.	Krishnachura or Gulmohar (Delonixregia) seeds
12.	Krishnachura or Gulmohar (Delonixregia) leaves
13.	Bean (Phaseolussp.) seeds
14.	Castor (Ricinuscommunis) seeds
15.	Castor (Ricinuscommunis) seed coats
16.	Lai (Brassica juncea) seeds
17.	Denga (Amaranthus spinosus) seeds
18.	Sandal wood (Santalumsp.) seeds
19.	Kanchan (Bauhinia acuminata) seeds
20.	Tishi(Linumusitatissimun) seeds
21.	Mahogany (Swieteniamehogoni) seed coats
22.	Crassocephalumcrepidioidesseed hairs
23.	Taal or Palmyra (Borassusflabellier) fibres
24.	Taal or Palmyra (Borassusflabellier) seeds
25.	Bladygrass (Imperatacylindrica) inflorescences
26.	Purol(Luffacylindrica) fibres
27.	Pajanelialongifolia seeds
28.	Daisy (Tridaxprocumbens) seed hairs
29.	Eulaliafastigiata inflorescences
30.	Ishabgul orPsyllium (Plantago sp.) seed husk

 Table 2.1 Precursors used for the nanomaterials synthesis

 31.	Papaya (Carica papaya) stems fibre
32.	Oroxylumindicumseeds
33.	Papery bracts of Paper flower (Bougainvillea spectabilis)
34.	Bontula (Bombax insigne)
35.	Thunga or Takpalong(Spinaciaoleracea) seed coats
36.	Betel nut (Areca catechu) fibres
37.	Blue-green alga, Scytonemaguyanense
38.	Green alga, Trentepohliaaurea
39.	Green alga, spirogyraneglecta
40.	Plant charcoal
41.	Waste bond paper
42.	Dalbergiasissoo leaves