2017/EVEN/12/31/AE-601(C)/024

B.Tech Even Semester (CBCS) Exam., April-2017

AGRICULTURAL ENGINEERING

(6th Semester)

Course No. : AE-601 (C)

(Thermal Operation in Food Processing)

Full Marks : 75Pass Marks : 30

Time : 3 hours

- Note: 1. Attempt one question from each Unit.
 - 2. Begin each answer in a new page.
 - 3. Answer parts of a question at a place.
 - 4. Assume reasonable data wherever required.
 - 5. The figures in the margin indicate full marks for the questions.

Unit—1

(a) Derive expression for calculating temperature and total amount of heat transfer from a body at any instant of time by using Newtonian cooling. Assume thermal resistance of body is zero.

J7**/1532**

(Turn Over)

(2)

- (b) Define Fourier law of heat transfer for steady state. Also find the rate of heat flow through a slab (assume as per requirement).
 3+2=5
- (a) What is critical thickness of insulation on a small diameter wire or pipe? Explain its physical significance and derive an expression for the same.
 - (b) Define the following : $1\frac{1}{2}\times4=6$
 - (i) Nusselt number
 - (ii) Prandtl number
 - (iii) Biot number
 - (iv) Unsteady state heat transfer

Unit—2

- **3.** (a) Derive an expression for log mean temperature difference (LMTD) of a heat exchanger. 7
 - (b) A simple counter flow heat exchanger operates under the following condition.
 Fluid—A, inlet and outlet temperatures 80 °C and 40 °C.
 Fluid–B, inlet and outlet temperatures 20 °C and 40 °C.
 The exchanger is cleaned, causing an

(Continued)

increase in the overall heat coefficient by 10% and inlet temperature of fluid B is changed to 30 °C. What will be the new outlet temperature of fluid A and of fluid B? Assume heat transfer coefficient and capacity rate are unaltered by temperature changes.

- **4.** (a) What is effectiveness of heat exchanger? Derive an expression for the effectiveness of a parallel flow heat exchanger. 2+6=8
 - (b) Nitrobenzene of 0.2 kg/sec is to be cooled from 400 K to 315 K by heating a stream of benzene from 305 K to 345 K. A tubular heat exchanger is available with a shell fitted with 165 tube (OD = 19 mm and ID = 15 mm) each 5 m long. What value of scale resistance on the outer surface of the inner tube could be allowed if these units were used?

The benzene side heat transfer coefficient h_i 1000 w / m²-k (which flow through the tube) and nitrobenzene side heat transfer coefficient h_0 750 w / m²-k, temperature correction factor 0 8, specific heat capacity of nitrobenzene 2380 J / kg-k.

J7**/1532**

7

8

UNIT—3

- **5.** (a) Discuss the following :
 - (i) D-value
 - (ii) Z-value
 - (iii) Arrhenius equation
 - (b) A suspension containing 3 10^5 spores of organism A having a D-value of 1.5 min at 121.1 °C and 8 10^6 spores of organism B having a D-value of 0.8 min at 121.1 °C is heated at a uniform constant temperature of 121.1 °C needed to obtain a probability of spoilage of 1/1000.
- **6.** (a) What is order of reaction? Describe zero order and first order reaction. $2+3\frac{1}{2}+3\frac{1}{2}=9$
 - (b) Write short notes on : $2 \times 3 = 6$
 - (i) Pasteurization
 - (ii) Sterilization
 - (iii) UHT processing

Unit—4

- 7. (a) Explain the drying mechanism of moisture removal in solid food.7
 - (b) Find an expression for the calculation of total drying time in solid food.
- J7**/1532**

 $3 \times 3 = 9$

6

(5)

- 8. (a) Explain the working function of a drum dryer with the help of a neat diagram.7
 - (b) Explain the freeze drying process and also mention the advantages of freeze drying system.

8

6

9

Unit—5

- **9.** (a) An evaporator is used to concentrate cane sugar. A feed of 100 kg/h of a solution containing 38% sugar is evaporated producing a 74% solution. Calculate the weight of solution produced and amount of water removed.
 - (b) Define water activity. Explain the method for measurement of water activity at high moisture content. 2+7=9
- **10.** (*a*) Write down the methods of operation of evaporators and calculation methods for single-effect evaporators.
 - (b) Calculate the water activity of a 50% sucrose solution. (Given : K = 2 7, molecular weight of sucrose is 342) 6

 $\star \star \star$