

(4)

- (b) Write the Flow chart and Algorithm to convert the temperature to Fahrenheit. 2+2=4
- (c) Write the Algorithm and Fortran77 Program to calculate the rate constant of a first-order reaction. 3+3=6

## Unit-V

9. (a) What are the terms involved in Schrodinger equation of a many body system? How Born-Openheimer approximation is used to solve the Schrodinger equation? Molecular geometry depends on the mass of the nuclei but not on the charge. Explain. 2+3+3=8
- (b) What is potential energy surface (PES). Draw and explain PES of H<sub>2</sub>O molecule considering its C<sub>2v</sub> symmetry. 6
10. (a) What are the objectives of computational chemistry? Explain how abinitio method is different from DFT. 2+3= 5
- (b) What are the basis sets? Explain the 6-31G basis sets? Calculate the number of basis function and primitives in the case of CH<sub>2</sub> molecule using 6-31G basis set. 2+3+4=9

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## PG (CBCS) EVEN SEMESTER EXAMINATION, 2023

## CHEMISTRY

4th Semester

Course No. : CHMCC - 401

## ( Analytical and Computational Chemistry )

Full Marks : 70

Pass Marks : 28

Time : 3 hours

*The figures in the margin indicate full marks for the questions*

(Answer five questions, selecting one from each unit)

## Unit-I

1. (a) In a chromatographic analysis of lemon oil a peak for limonene has a retention time of 8.36 min with a baseline width of 0.96 min.  $\gamma$ -Terpinene elutes at 9.54 min with a baseline width of 0.64 min. The column void time is 0.31 min. What is the resolution between the peaks? Calculate capacity factor of Limonene and  $\gamma$ - Terpinene and selective factor for Limonene and  $\gamma$ - Terpinene. 4
- (b) Provide the structures of silica and alumina and explain their role as stationary phase. 4
- (c) Give a comparison between normal phase chromatography and reverse phase chromatography. 3
- (d) State basic principle and some application of ion exchange chromatography. 3

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(2)

- 2) (a) Explain the working principle of GC with suitable schematic diagram and mention its application. 6+2 = 8
- (b) What are the main difference between HPLC and GC. 3
- (c) Establish Van-Deemter equation with illustration of band broadening process. 3

#### Unit-II

3. (a) Define polarography. Mention the two important steps for any electrolysis and discuss. 1+3=4
- (b) Derive the relation between limiting diffusion current and depolarizer concentration in polarography where DME is used as working electrode. 4
- (c) Draw and explain the cyclic voltammogram for  $K_3[Fe(CN)_6]$  (Solvent  $H_2O$ ; Supporting electrolyte,  $KNO_3$ ; scan rate 50 mV/s, Potential window +0.8 to -0.2 V) 1+3 = 4
- (d) Calculate the number of electrons transferred and formal potential for a redox couple in cyclic voltammetry, where  $E_{pc} = 0.555V$  and  $E_{pa} = 0.492$ . 2
4. (a) Briefly explain the role of supporting electrolyte in polarographic analysis by taking a suitable example. 2
- (b) Write a short note on Dropping Mercury electrode (DME). 3
- (c) Write down the fundamental principle of amperometric titration. By taking suitable examples, briefly discuss the most common types of curve encountered in amperometric titrations. 2+4= 6

(3)

- (d) Derive the relation between current vs. time in the controlled potential electrolysis. 3

#### Unit-III

5. (a) What are the factors that affect the result of a thermogravimetrythermogram? 4
- (b) How to calculate mass loss and determine the offset and onset temperature of TG thermogram. 3
- (c) Discuss the advantage of DTA over TG. 2
- (d) What is differential scanning calorimetry? Discuss the types of differential scanning calorimetry. 1+4 =5
6. (a) What is the basis of neutron activation analysis(NAA)? Explain the terms PGNAA and DGNAA. Explain, how the NAA technique be used in archaeology and forensic investigation. 2+2+4=8
- (b) Briefly describe the method of determination of vanadium in lubricating oil by AAS. 3
- (c) What are the strength and limitation of XRF? 3

#### Unit-IV

7. (a) What are the hardware components of a computer? Discuss briefly about them. 3+5=8
- (b) Write the Flow chart and Fortran77 Program to calculate the pressure of a real gasusing van der Waal's equation. 3+3=6
8. (a) What is a Flow chart? Why do we need a flow chart? 1+3=4

( Turn Over )