III. Short answers (Answer seven out of nine questions) 5x7=35

 What are spectrophotometric titrations. Describe its instrumentation with neat and labelled diagram. 1+4=5

4

- 2. Explain the various factors that affect separation during ion exchange chromatography. 5
- 3. What is Beer-Lambert law? Mention the limitations of Beer-Lambert law. 2+3=5
- 4. Write a note on theory and applications of affinity chromatography. 5
- 5. Describe the various factors affecting the scattering of light. 5
- 6. Explain the process of electroosmotic flow in capillary electrophoresis (Give diagram). 5
- 7. Describe the instrumentation of flame photometer along with neat and labelled diagram. 5
- 8. Discuss the principle of atomic absorption spectroscopy. 5
- Enumerate the detectors used in IR spectroscopy & describe any two in detail.

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2023/SEM/ODD/BP-701T/008

UG Odd Semester (CBCS) Examination, March 2023

PHARMACEUTICAL SCIENCES

(7th Semester)

Course No: BP 701T

(Instrumental Methods of Analysis-Theory)

Full Marks: 75

Time: 3 Hours

The figures in the margin indicate full marks for the questions

I (A). Multiple choice questions

1x10=10

1. The energies required for the following transitions in increasing order:

- 2. Cross-linked dextran having porous 3D structure are specifically used in
 - (a) Affinity chromatography
 - (b) Molecular Seive chromatography
 - (c) Gel electrophoresis
 - (d) Ion exchange chromatography
- 3. Which of the following flame is used for analyzing refractory elements?
 - (a) Oxygen-hydrogen
 - (b) Air-acetylene
 - (c) Nitrous oxide-acetylene
 - (d) Air-propane

- 4. SDS-PAGE is a technique widely used in the separation of
 - (a) Carbohydrates
 - (b) Lipids
 - (c) Proteins
 - (d) Nucleic acid
- 5. Which of the following is used for the determination of high concentration suspension?
 - (a) Refractometry
 - (b) Nephelometry
 - (c) Turbidimetry
 - (d) Both (b) and (c)
- 6. Immobilized metal ion affinity chromatography is based on the specific coordinate covalent bond between metals and
 - (a) Alanine
 - (b) Histidine
 - (c) Glycine
 - (d) Histamine
- 7. Amine shows characteristic N-H stretch at
 - (a) 3650 3200 cm-1
 - (b) 3500 3300 cm-1
 - (c) 3000 3300 cm-1
 - (d) None of the above
- 8. For better chromatographic separations in gas chromatography, barbiturates are derivatized with the aid of
 - (a) Tripropylammonium hydroxide
 - (b) Trimethylanilinium hydroxide
 - (c) Triethylanilinium hydroxide
 - (d) Triethylammonium hydroxide
- 9. Quantum yield of highly fluorescent substances is(a) 0
 - (b) Near to 1

- (c) Less than 1
- (d) None of the above
- 10. For increased flexibility, capillaries used in capillary electrophoresis are coated with a polymer
 - (a) Polystyrene-Isoprene
 - (b) Polyacrylamide
 - (c) Polyimide
 - (d) Polyisoprene

I (B). Objective type (Answer the following in brief)

2x5=10

- 1. Define chromophore and auxochrome with examples.
- 2. What is the total number of theoretical plates if the HETP is 0.7 mm and the length of the packed column is 15 cm?
- 3. Mention the functions of nebulizer in atomic absorption spectrometer.
- 4. Write a note on the Nernst glower.
- 5. Mention two detectors used in HPLC.

II. Long answers (Answer two out of three questions)

10x2=20

- 1. Describe the factors affecting fluorescence intensity.
- 2. Discuss the theory, instrumentaion (including detectors) and applications of gas chromatography. Give necessary diagram.
- 3. Explain the principle, development technique (give necessary diagrams) and application of column adsorption chromatography.