2016/ODD/07/20/BSCP-502/ BSMP-502/460

UG Odd Semester (CBCS) Exam., December-2016

B.Sc (Honours) B.Ed

(5th Semester)

Course No. : BSCP-502/BSMP-502

Time : 2 hours

The figures in the margin indicate full marks for the questions

Candidates are to answer *either* BSCP–502 *or* BSMP–502

CHEMISTRY

(Pass)

Course No. : BSCP-502

(General Chemistry)

Full Marks : 50 Pass Marks : 20

- **1.** (*a*) Show diagrammatically the crystal-field splitting of *d*-orbitals in an octahedral complex.
 - (b) What is spectrochemical series? Write one of its uses in coordination chemistry.
 - *(c)* Write one limitation of valence-bond theory.
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2

4

2

(2)

(d) Define crystal-field stabilization energy (CFSE). 2

OR

- **2.** (*a*) Write the differences between bonding and antibonding molecular orbitals. 2
 - (b) Show diagrammatically the molecular orbital diagram of NO molecule and calculate its bond order.
 - (c) Using VSEPR theory, predict the shapes of XeF_4 and H_2O molecules. 4
- **3.** (a) Arrange the compounds in order of increasing acidity and explain the reasons : 2×3=6
 - (i) H_2O , CH_4 , NH_3 , HF
 - (ii) Ethylene, Ethane, Acetylene
 - (iii) H_3CCOOH , CH_2FCOOH , $CH_2CICOOH$, CH_3CH_2COOH
 - (b) Arrange the compounds in order of increasing solubility with reason : 2

 $\begin{array}{c} CH_2OH \\ | \\ CHOH \\ | \\ H_2OH \end{array}, CH_3CH_2OH, \begin{array}{c} CH_2OH \\ | \\ CH_2OH \end{array}$

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(Continued)

(3)

(c) Explain why chloroform $CHCl_3$ is polar but carbon tetrachloride CCl_4 is nonpolar.

OR

| (a) | Write the Knorr synthesis of pyrrole with mechanism. | 4 |
|-----|-----------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------------|
| (b) | Why is pyridine more basic than pyrrole? | 2 |
| (c) | Write the Fischer indole synthesis with mechanism. | 4 |
| (a) | Write the mechanism of nitration of benzene. | 3 |
| (b) | Write the structure of Gly-Ala peptide bond. | 2 |
| (c) | Write the mechanism of Strecker synthesis of amino acid. | 3 |
| (d) | What is zwitterion? Write the structure of zwitterion of alanine. | 2 |
| | (b) (c) (a) (b) (c) | (b) Why is pyridine more basic than pyrrole? (c) Write the Fischer indole synthesis with mechanism. (a) Write the mechanism of nitration of benzene. (b) Write the structure of Gly-Ala peptide bond. (c) Write the mechanism of Strecker synthesis of amino acid. (d) What is zwitterion? Write the structure of |

OR

6. (a) Complete the following reactions : $2 \times 2 = 4$

$$(i) \quad \begin{array}{c} & O \\ \parallel \\ H_3C - C - CH_3 \end{array} \xrightarrow{1) H_3C - Mg - I} \\ \hline & 2) H_2O/H^+ \end{array}?$$

(*ii*) H₃C—C≡N
$$\xrightarrow{1)$$
 H₃C-Mg-I
2) H₂O/H⁺ ?

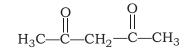
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(Turn Over)

2

(4)

- (b) Prepare cyclobutane carboxylic acid from diethyl malonate (DEM).3
- (c) Prepare



from suitable active methylene compound.

3

- **7.** (a) Draw and explain the phase diagram of H_2O system. 4
 - (b) The value of equilibrium constant K_P for the water gas reaction
 - CO (g) $H_2O(g) \rightleftharpoons CO_2(g) H_2(g)$ is 1 06 10⁵ at 25 °C. Calculate G for the reaction. 4
 - (c) Write Le Chatelier's principle and its application. 2

OR

- **8.** (a) Define colligative property with example. 2
 - (b) Write the Raoult's law and its mathematical form. 2
 - (c) Define with example the Hardy-Schulze rule. 2

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(Continued)

(5)

- (d) A solution of 12.5 g of urea in 170 g of water gave boiling point elevation of 0 63 K. Calculate the molar mass of urea. K_b 0 52 K kg mol⁻¹.
- **9.** Explain the following terms : $2\frac{1}{2}\times4=10$
 - (a) Fluorescence
 - (b) Phosphorescence
 - (c) Quantum yield
 - (d) Non-radiative transitions

OR

| 10. | (a) | Write the expression of rotational energy in the <i>J</i> th level. | 2 |
|-----|-----|-----------------------------------------------------------------------------------|---|
| | (b) | What is reduced mass? | 2 |
| | (c) | Show that rotational lines are equally spaced by a constant spacing $(2B)$ during | |
| | | transition. | 4 |
| | (d) | What do you mean by zero point energy? | 2 |
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MATHEMATICS

(Pass)

Course No. : BSMP-502

(Coordinate Geometry and Linear Programming)

Full Marks : 70 Pass Marks : 28

- **1.** (a) Solve the equation :
 - 4*y* 12*y* 9*y* 0
 - (b) Define second-order linear differential equation. 7

OR

| 2. | (a) | Find the equation of tangent of the circle | | | | | | | | | |
|----|-----|--------------------------------------------|-------|------------------|------------|----|---|----|------|----|--|
| | | x^2 | y^2 | 6 <i>x</i> | 4 <i>y</i> | 12 | 0 | W | hich | is | |
| | | parallel t | | to the line $4x$ | | 3у | 5 | 0. | | 7 | |

- (b) Find the pole of the line 3x 5y 17 with respect to the circle x^2 y^2 4x 6y 9 0.
 - 7

7

3. (a) If a, b, h are not all zero and h^2 ab, then prove that $ax^2 2hxy by^2 0$ represents a pair of straight lines passing through origin. 7

4

(7)

(b) Show that the straight lines represented by $(x \ 2a)^2 \ 3y^2 \ 0$ and $x \ a$ form an equilateral triangle.

OR

- **4.** (a) Determine the vertex and axis of symmetry for the graph of each equation : 4×2=8
 - (*i*) $y 2(x 4)^2 3$ (*ii*) $y x^2 4x 8$
 - (b) Find the equation of vertical parabola with a vertex (2, 6) and passing through the point (1, 4).
- 5. (a) Define equation of tangent and normal of a curve.6
 - (b) Find the equation of the tangent to $f(x) = x^3 = 3x^2 = x = 1$ at the point, where x = 3. OR

- **6.** (a) Define plane and deduct the equation of plane passing through (x_1, y_1, z_1) . 7
 - (b) Find the equation of the plane passing through points (2, 2, 1), (1, 2, 3) and parallel to x-axis.7

7

6

8

(8)

| 7. | (a) | Explain linear programming model. | 7 | | | | | | | |
|-----|-----|-----------------------------------------------------------|---|--|--|--|--|--|--|--|
| | (b) | What are the important assumptions of linear programming? | 7 | | | | | | | |
| | | OR | | | | | | | | |
| 8. | (a) | Define dual problem. | 6 | | | | | | | |
| | (b) | Explain the relationship between primal and dual problem. | 8 | | | | | | | |
| 9. | (a) | Define dual simplex algorithm. | 7 | | | | | | | |
| | (b) | Write a note on game theory. | 7 | | | | | | | |
| OR | | | | | | | | | | |
| 10. | (a) | What is test of optimality? | 6 | | | | | | | |
| | (b) | Explain explicit tests of optimality in a Hilbert space. | 8 | | | | | | | |

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